

Monoclinic, $P2_1/c$
 $a = 20.5417 (6)$ Å
 $b = 7.1358 (2)$ Å
 $c = 6.3402 (2)$ Å
 $\beta = 93.632 (2)^\circ$
 $V = 927.49 (5)$ Å³

$Z = 2$
Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 100 (2)$ K
 $0.40 \times 0.25 \times 0.10$ mm

Biphenyl-4-carbaldehyde azine

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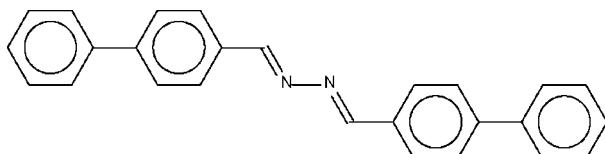
Received 18 November 2008; accepted 19 November 2008

Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(C-C) = 0.002$ Å;
 R factor = 0.042; wR factor = 0.127; data-to-parameter ratio = 16.6.

The complete molecule of the title compound, C₂₆H₂₀N₂, is generated by crystallographic inversion symmetry. The terminal phenyl ring is twisted by 19.2 (1) $^\circ$ with respect to the adjacent phenylene ring.

Related literature

For the synthesis, see: Malkes & Timchenko (1961). For biological evaluation, see: Cremlyn *et al.* (1991). The compound is a formylating agent for aromatic compounds; see: Kantlehner *et al.* (2004). When treated with cerium ammonium nitrate, the aldehyde is regenerated; see Giurg & Mlochowski (1999).



Experimental

Crystal data

C₂₆H₂₀N₂

$M_r = 360.44$

Data collection

Bruker SMART APEX CCD diffractometer
Absorption correction: none
6044 measured reflections

2104 independent reflections
1607 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.025$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.042$
 $wR(F^2) = 0.127$
 $S = 1.05$
2104 reflections

127 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.31$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.21$ e Å⁻³

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: X-SEED (Barbour, 2001); software used to prepare material for publication: publCIF (Westrip, 2008).

We thank the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2856).

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supplementary materials

Acta Cryst. (2008). E64, o2444 [doi:10.1107/S1600536808038622]

Biphenyl-4-carbaldehyde azine

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Comment

The complete molecule of the title compound, (I) is generated by crystallographic inversion symmetry (Fig. 1). The terminal phenyl ring is twisted by 19.2 (1)° with respect to the phenylene ring.

Experimental

4-Phenyl benzaldehyde (0.72 g, 4 mmol) and 80% hydrazine hydrate (0.10 g, 2 mmol) were heated in ethanol (25 ml) for 1 h. The resulting product was filtered and washed with ethanol and then recrystallized from hexane to yield yellow prisms of (I).

Refinement

The H atoms were placed in calculated positions (C—H = 0.95 Å) and refined as riding with $U(\text{H}) = 1.2U(\text{C})$.

Figures

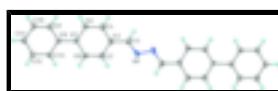


Fig. 1. The molecular structure of (I) with atoms shown at the 70% probability level. Hydrogen atoms are drawn as spheres of arbitrary radius. The unlabelled atoms are generated by the symmetry operation (1-x, 1-y, 1-z).

Biphenyl-4-carbaldehyde azine

Crystal data

$\text{C}_{26}\text{H}_{20}\text{N}_2$	$F_{000} = 380$
$M_r = 360.44$	$D_x = 1.291 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
Hall symbol: -P 2ybc	$\lambda = 0.71073 \text{ \AA}$
$a = 20.5417(6) \text{ \AA}$	Cell parameters from 1869 reflections
$b = 7.1358(2) \text{ \AA}$	$\theta = 2.9\text{--}26.2^\circ$
$c = 6.3402(2) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$\beta = 93.632(2)^\circ$	$T = 100(2) \text{ K}$
$V = 927.49(5) \text{ \AA}^3$	Prism, yellow
$Z = 2$	$0.40 \times 0.25 \times 0.10 \text{ mm}$

Data collection

Bruker SMART APEX CCD diffractometer 1607 reflections with $I > 2\sigma(I)$

supplementary materials

Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.025$
Monochromator: graphite	$\theta_{\text{max}} = 27.5^\circ$
$T = 100(2)$ K	$\theta_{\text{min}} = 1.0^\circ$
ω scans	$h = -25 \rightarrow 26$
Absorption correction: None	$k = -8 \rightarrow 9$
6044 measured reflections	$l = -8 \rightarrow 8$
2104 independent reflections	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.042$	H-atom parameters constrained
$wR(F^2) = 0.127$	$w = 1/[\sigma^2(F_o^2) + (0.0641P)^2 + 0.2493P]$ where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.06$	$(\Delta/\sigma)_{\text{max}} = 0.001$
2104 reflections	$\Delta\rho_{\text{max}} = 0.31 \text{ e \AA}^{-3}$
127 parameters	$\Delta\rho_{\text{min}} = -0.21 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.46654 (6)	0.48506 (18)	0.47390 (19)	0.0260 (3)
C1	0.45139 (7)	0.5211 (2)	0.2795 (2)	0.0229 (3)
H1	0.4847	0.5598	0.1916	0.027*
C2	0.38464 (7)	0.50463 (19)	0.1880 (2)	0.0202 (3)
C3	0.37032 (7)	0.5594 (2)	-0.0206 (2)	0.0218 (3)
H3	0.4045	0.6006	-0.1034	0.026*
C4	0.30685 (7)	0.5548 (2)	-0.1095 (2)	0.0205 (3)
H4	0.2983	0.5927	-0.2522	0.025*
C5	0.25534 (6)	0.49541 (19)	0.0078 (2)	0.0169 (3)
C6	0.27039 (7)	0.43846 (19)	0.2177 (2)	0.0198 (3)
H6	0.2363	0.3972	0.3008	0.024*
C7	0.33352 (7)	0.4412 (2)	0.3054 (2)	0.0212 (3)
H7	0.3424	0.3998	0.4467	0.025*
C8	0.18693 (6)	0.49529 (18)	-0.08434 (19)	0.0170 (3)
C9	0.16848 (7)	0.60293 (19)	-0.2631 (2)	0.0202 (3)

H9	0.2005	0.6745	-0.3287	0.024*
C10	0.10445 (7)	0.6072 (2)	-0.3461 (2)	0.0215 (3)
H10	0.0932	0.6811	-0.4676	0.026*
C11	0.05681 (7)	0.5041 (2)	-0.2529 (2)	0.0200 (3)
H11	0.0128	0.5083	-0.3084	0.024*
C12	0.07425 (6)	0.39455 (19)	-0.0772 (2)	0.0197 (3)
H12	0.0420	0.3224	-0.0132	0.024*
C13	0.13839 (6)	0.38970 (19)	0.0054 (2)	0.0186 (3)
H13	0.1496	0.3133	0.1249	0.022*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0154 (6)	0.0334 (7)	0.0285 (7)	-0.0007 (5)	-0.0034 (5)	-0.0004 (5)
C1	0.0174 (7)	0.0249 (8)	0.0261 (7)	0.0009 (5)	-0.0003 (5)	-0.0009 (6)
C2	0.0171 (7)	0.0195 (7)	0.0235 (7)	0.0013 (5)	-0.0018 (5)	-0.0020 (5)
C3	0.0181 (7)	0.0244 (7)	0.0231 (7)	-0.0002 (5)	0.0028 (5)	0.0015 (5)
C4	0.0205 (7)	0.0229 (7)	0.0178 (6)	0.0009 (5)	0.0002 (5)	0.0015 (5)
C5	0.0168 (7)	0.0149 (6)	0.0187 (6)	0.0012 (5)	-0.0017 (5)	-0.0015 (5)
C6	0.0188 (7)	0.0209 (7)	0.0196 (6)	-0.0007 (5)	0.0013 (5)	0.0015 (5)
C7	0.0221 (7)	0.0225 (7)	0.0186 (6)	0.0005 (5)	-0.0020 (5)	0.0007 (5)
C8	0.0185 (7)	0.0164 (6)	0.0160 (6)	0.0012 (5)	-0.0010 (5)	-0.0028 (5)
C9	0.0203 (7)	0.0201 (7)	0.0200 (6)	-0.0024 (5)	0.0000 (5)	0.0025 (5)
C10	0.0245 (7)	0.0213 (7)	0.0182 (6)	0.0011 (5)	-0.0036 (5)	0.0020 (5)
C11	0.0170 (7)	0.0227 (7)	0.0197 (6)	0.0018 (5)	-0.0037 (5)	-0.0036 (5)
C12	0.0185 (7)	0.0210 (7)	0.0196 (6)	-0.0016 (5)	0.0022 (5)	-0.0010 (5)
C13	0.0195 (7)	0.0191 (7)	0.0169 (6)	0.0008 (5)	-0.0004 (5)	0.0010 (5)

Geometric parameters (\AA , $^\circ$)

N1—C1	1.2784 (19)	C6—H6	0.9500
N1—N1 ⁱ	1.410 (2)	C7—H7	0.9500
C1—C2	1.4592 (18)	C8—C13	1.3989 (18)
C1—H1	0.9500	C8—C9	1.4012 (18)
C2—C3	1.3927 (18)	C9—C10	1.3858 (19)
C2—C7	1.4006 (19)	C9—H9	0.9500
C3—C4	1.3874 (18)	C10—C11	1.3863 (19)
C3—H3	0.9500	C10—H10	0.9500
C4—C5	1.3969 (19)	C11—C12	1.3888 (19)
C4—H4	0.9500	C11—H11	0.9500
C5—C6	1.4073 (18)	C12—C13	1.3870 (18)
C5—C8	1.4873 (17)	C12—H12	0.9500
C6—C7	1.3784 (18)	C13—H13	0.9500
C1—N1—N1 ⁱ	111.73 (15)	C6—C7—H7	119.7
N1—C1—C2	122.17 (13)	C2—C7—H7	119.7
N1—C1—H1	118.9	C13—C8—C9	117.44 (12)
C2—C1—H1	118.9	C13—C8—C5	121.38 (11)
C3—C2—C7	118.38 (12)	C9—C8—C5	121.17 (12)

supplementary materials

C3—C2—C1	119.45 (13)	C10—C9—C8	121.38 (12)
C7—C2—C1	122.14 (12)	C10—C9—H9	119.3
C4—C3—C2	121.03 (12)	C8—C9—H9	119.3
C4—C3—H3	119.5	C9—C10—C11	120.34 (12)
C2—C3—H3	119.5	C9—C10—H10	119.8
C3—C4—C5	121.01 (12)	C11—C10—H10	119.8
C3—C4—H4	119.5	C10—C11—C12	119.17 (12)
C5—C4—H4	119.5	C10—C11—H11	120.4
C4—C5—C6	117.55 (12)	C12—C11—H11	120.4
C4—C5—C8	121.34 (11)	C13—C12—C11	120.49 (13)
C6—C5—C8	121.11 (12)	C13—C12—H12	119.8
C7—C6—C5	121.48 (12)	C11—C12—H12	119.8
C7—C6—H6	119.3	C12—C13—C8	121.16 (12)
C5—C6—H6	119.3	C12—C13—H13	119.4
C6—C7—C2	120.53 (12)	C8—C13—H13	119.4
N1 ⁱ —N1—C1—C2	-178.70 (14)	C4—C5—C8—C13	161.70 (13)
N1—C1—C2—C3	175.14 (14)	C6—C5—C8—C13	-19.36 (19)
N1—C1—C2—C7	-2.6 (2)	C4—C5—C8—C9	-19.07 (19)
C7—C2—C3—C4	1.2 (2)	C6—C5—C8—C9	159.86 (13)
C1—C2—C3—C4	-176.66 (13)	C13—C8—C9—C10	0.99 (19)
C2—C3—C4—C5	0.1 (2)	C5—C8—C9—C10	-178.27 (12)
C3—C4—C5—C6	-0.8 (2)	C8—C9—C10—C11	0.1 (2)
C3—C4—C5—C8	178.21 (13)	C9—C10—C11—C12	-0.9 (2)
C4—C5—C6—C7	0.1 (2)	C10—C11—C12—C13	0.7 (2)
C8—C5—C6—C7	-178.86 (12)	C11—C12—C13—C8	0.5 (2)
C5—C6—C7—C2	1.2 (2)	C9—C8—C13—C12	-1.26 (19)
C3—C2—C7—C6	-1.8 (2)	C5—C8—C13—C12	177.99 (12)
C1—C2—C7—C6	175.96 (13)		

Symmetry codes: (i) $-x+1, -y+1, -z+1$.

Fig. 1

